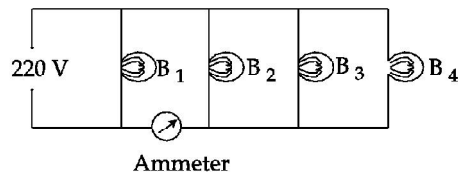


# JEE-MAIN & AIEEE

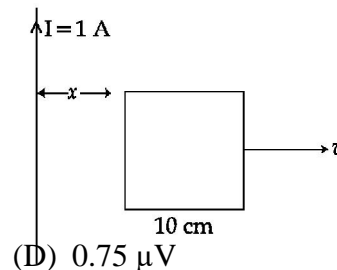
## ONLINE EXAM TEST PAPERS OF 2014 (PAPER-4)

- Q.1 The velocity of water in a river is 18 km/hr near the surface. If the river is 5 m deep, find the shearing stress between the horizontal layers of water. The co-efficient of viscosity of water =  $10^{-2}$  poise.  
 (A)  $10^{-1}$  N/m<sup>2</sup> (B)  $10^{-4}$  N/m<sup>2</sup> (C)  $10^{-2}$  N/m<sup>2</sup> (D)  $10^{-3}$  N/m<sup>2</sup>
- Q.2 Four bulbs B<sub>1</sub>, B<sub>2</sub>, B<sub>3</sub> and B<sub>4</sub> of 100 W each are connected to 220 V main as shown in the figure. The reading in an ideal ammeter will be:



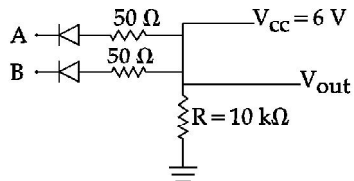
- (A) 1.35 A (B) 0.45 A (C) 1.80 A (D) 0.90 A
- Q.3 In a Young's double slit experiment, the distance between the two identical slits is 6.1 times larger than the slit width. Then the number of intensity maxima observed within the central maximum of the single slit diffraction pattern is :  
 (A) 12 (B) 24 (C) 3 (D) 6
- Q.4 A body is in simple harmonic motion with time period half second ( $T = 0.5$  s) and amplitude one cm ( $A = 1$  cm). Find the average velocity in the interval in which it moves from equilibrium position to half of its amplitude.  
 (A) 6 cm/s (B) 4 cm/s (C) 12 cm/s (D) 16 cm/s
- Q.5 A bullet loses  $\left(\frac{1}{n}\right)^{\text{th}}$  of its velocity passing through one plank. The number of such planks that are required to stop the bullet can be:  
 (A) n (B)  $\frac{n^2}{2n-1}$  (C) Infinite (D)  $\frac{2n^2}{n-1}$
- Q.6 Long range radio transmission is possible when the radiowaves are reflected from the ionosphere. For this to happen the frequency of the radiowaves must be in the range :  
 (A) 150 - 500 kHz (B) 8 - 25 MHz (C) 80 - 150 MHz (D) 1-3 MHz
- Q.7 A ray of light is incident from a denser to a rarer medium. The critical angle for total internal reflection is  $\theta_{ic}$  and the Brewster's angle of incidence is  $\theta_{ib}$ , such that  $\sin\theta_{ic}/\sin\theta_{ib} = \eta = 1.28$ . The relative refractive index of the two media is :  
 (A) 0.4 (B) 0.9 (C) 0.8 (D) 0.2
- Q.8 If denote microwaves, X rays, infrared, gamma rays, ultra-violet, radio waves and visible parts of the electromagnetic spectrum by M, X, I, G, U, R and V, the following is the arrangement in ascending order of wavelength :  
 (A) M, R, V, X, U, G and I (B) R, M, I, V, U, X and G  
 (C) G, X, U, V, I, M and R (D) I, M, R, U, V, X and G
- Q.9 A piece of wood from a recently cut tree shows 20 decays per minute. A wooden piece of same size placed in a museum (obtained from a tree cut many years back) shows 2 decays per minute. If half life of  $C^{14}$  is 5730 years, then age of the wooden piece placed in the museum is approximately :  
 (A) 19039 years (B) 10439 years (C) 39049 years (D) 13094 years

Q.10 A square frame of side 10 cm and a long straight wire carrying current 1 A are in the plane of the paper. Starting from close to the wire, the frame moves towards the right with a constant speed of  $10 \text{ ms}^{-1}$  (see figure). The e.m.f induced at the time the left arm of the frame is at  $x = 10 \text{ cm}$  from the wire is :



- (A)  $1 \mu\text{V}$                       (B)  $0.5 \mu\text{V}$                       (C)  $2 \mu\text{V}$                       (D)  $0.75 \mu\text{V}$

Q.11



Given : A and B are input terminals.

Logic 1 = > 5 V

Logic 0 = < 1 V

Which logic gate operation, the following circuit does ?

- (A) OR Gate                      (B) XOR Gate                      (C) AND Gate                      (D) NOR Gate

Q.12 The gravitational field in a region is given by  $\vec{g} = 5\text{N/kg} \hat{i} + 12\text{N/kg} \hat{j}$ . The change in the gravitational potential energy of a particle of mass 2 kg when it is taken from the origin to a point (7 m, -3 m) is:

- (A) 1 J                      (B)  $13\sqrt{58}$  J                      (C) -71 J                      (D) 71 J

Q.13 Match **List - I** (Event) with **List - II** (Order of the time interval for happening of the event) and select the correct option from the options given below the lists.

**List - I**

**List - II**

- |                                |                            |
|--------------------------------|----------------------------|
| (a) Rotation period of earth   | (i) $10^5 \text{ s}$       |
| (b) Revolution period of earth | (ii) $10^7 \text{ s}$      |
| (c) Period of a light wave     | (iii) $10^{-15} \text{ s}$ |
| (d) Period of a sound wave     | (iv) $10^{-3} \text{ s}$   |
- (A) (a)-(ii), (b)-(i), (c)-(iv), (d)-(iii)                      (B) (a)-(i), (b)-(ii), (c)-(iv), (d)-(iii)  
 (C) (a)-(ii), (b)-(i), (c)-(iii), (d)-(iv)                      (D) (a)-(i), (b)-(ii), (c)-(iii), (d)-(iv)

Q.14 Match **List-I** (Experiment performed) with **List II** (Phenomena discovered/associated) and select the correct option from the options given below the lists:

List-I		List-II	
(a)	Davission and Germer experiment	(i)	Wave nature of electrons
(b)	Millikan's oil drop experiment	(ii)	Chage of an electron
(c)	Rutherford experiment	(iii)	Quantisation of energy levels
(d)	Franck Hertz experiment	(iv)	Existence of nucleus

- (A) (a)-(i), (b)-(ii), (c)-(iii), (d)-(iv)                      (B) (a)-(i), (b)-(ii), (c)-(iv), (d)-(iii)  
 (C) (a)-(iv), (b)-(iii), (c)-(ii), (d)-(i)                      (D) (a)-(iii), (b)-(iv), (c)-(i), (d)-(ii)

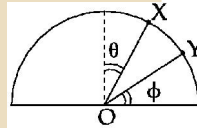
Q.15 A heavy box is to be dragged along a rough horizontal floor. To do so, person A pushes it at an angle  $30^\circ$  from the horizontal and requires a minimum force  $F_A$ , while person B pulls the box at an angle  $60^\circ$  from the horizontal and needs minimum force  $F_B$ . If the coefficient of friction between the box and the floor is  $\frac{\sqrt{3}}{5}$ , the ratio  $\frac{F_A}{F_B}$  is

- (A)  $\sqrt{\frac{3}{2}}$  (B)  $\frac{2}{\sqrt{3}}$  (C)  $\frac{5}{\sqrt{3}}$  (D)  $\sqrt{3}$

Q.16 A gas is compressed from a volume of  $2 \text{ m}^3$  to a volume of  $1 \text{ m}^3$  at a constant pressure of  $100 \text{ N/m}^2$ . Then it is heated at constant volume by supplying  $150 \text{ J}$  of energy. As a result, the internal energy of the gas:

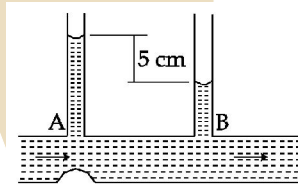
- (A) Increases by  $250 \text{ J}$  (B) Decreases by  $50 \text{ J}$   
 (C) Decreases by  $250 \text{ J}$  (D) Increases by  $50 \text{ J}$

Q.17 A particle is released on a vertical smooth semicircular track from point X so that OX makes angle  $\theta$  from the vertical (see figure). The normal reaction of the track on the particle vanishes at point Y where OY makes angle  $\phi$  with the horizontal. Then:



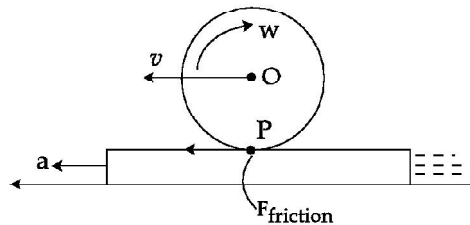
- (A)  $\sin\phi = \cos\theta$  (B)  $\sin\phi = \frac{1}{2} \cos\theta$  (C)  $\sin\phi = \frac{3}{4} \cos\theta$  (D)  $\sin\phi = \frac{2}{3} \cos\theta$

Q.18 In the diagram shown, the difference in the two tubes of the manometer is  $5 \text{ cm}$ , the cross section of the tube at A and B is  $6 \text{ mm}^2$  and  $10 \text{ mm}^2$  respectively. The rate at which water flows through the tube is ( $g = 10 \text{ ms}^{-2}$ )



- (A)  $10.0 \text{ cc/s}$  (B)  $7.5 \text{ cc/s}$  (C)  $8.0 \text{ cc/s}$  (D)  $12.5 \text{ cc/s}$

Q.19 Consider a cylinder of mass  $M$  resting on a rough horizontal rug that is pulled out from under it with acceleration 'a' perpendicular to the axis of the cylinder. What is  $F_{\text{friction}}$  at point P? It is assumed that the cylinder does not slip.



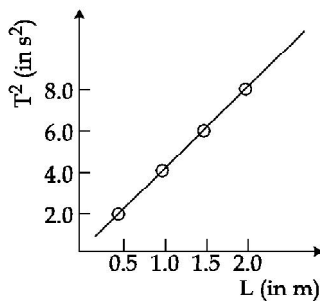
- (A)  $Mg$  (B)  $\frac{Ma}{3}$  (C)  $Ma$  (D)  $\frac{Ma}{2}$

Q.20 A large number of liquid drops each of radius  $r$  coalesce to form a single drop of radius  $R$ . The energy released in the process is converted into kinetic energy of the big drop so formed. The speed of the big drop is (given surface tension of liquid  $T$ , density  $\rho$ )

- (A)  $\sqrt{\frac{4T}{\rho} \left( \frac{1}{r} - \frac{1}{R} \right)}$  (B)  $\sqrt{\frac{T}{\rho} \left( \frac{1}{r} - \frac{1}{R} \right)}$  (C)  $\sqrt{\frac{2T}{\rho} \left( \frac{1}{r} - \frac{1}{R} \right)}$  (D)  $\sqrt{\frac{6T}{\rho} \left( \frac{1}{r} - \frac{1}{R} \right)}$

- Q.21 The gap between the plates of a parallel plate capacitor of area  $A$  and distance between plates  $d$ , is filled with a dielectric whose permittivity varies linearly from  $\epsilon_1$  at one plate to  $\epsilon_2$  at the other. The capacitance of capacitor is:  
 (A)  $\epsilon_0 (\epsilon_2 + \epsilon_1) A/2d$  (B)  $\epsilon_0 A/[d \ln(\epsilon_2/\epsilon_1)]$  (C)  $\epsilon_0 (\epsilon_1 + \epsilon_2) A/d$  (D)  $\epsilon_0 (\epsilon_2 - \epsilon_1) A/[d \ln(\epsilon_2/\epsilon_1)]$
- Q.22 A gas molecule of mass  $M$  at the surface of the Earth has kinetic energy equivalent to  $0^\circ\text{C}$ . If it were to go up straight without colliding with any other molecules, how high it would rise? Assume that the height attained is much less than radius of the earth. ( $k_B$  is Boltzmann constant)  
 (A)  $\frac{819k_B}{2Mg}$  (B)  $\frac{546k_B}{3Mg}$  (C) 0 (D)  $\frac{273k_B}{2Mg}$
- Q.23 The electric field in a region of space is given by,  $\vec{E} = E_0\hat{i} + 2E_0\hat{j}$  where  $E_0 = 100 \text{ N/C}$ . The flux of this field through a circular surface of radius  $0.02 \text{ m}$  parallel to the  $Y-Z$  plane is nearly  
 (A)  $0.02 \text{ Nm}^2/\text{C}$  (B)  $0.125 \text{ Nm}^2/\text{C}$  (C)  $3.14 \text{ Nm}^2/\text{C}$  (D)  $0.005 \text{ Nm}^2/\text{C}$
- Q.24 A ball of mass  $160 \text{ g}$  is thrown up at an angle of  $60^\circ$  to the horizontal at a speed of  $10 \text{ ms}^{-1}$ . The angular momentum of the ball at the highest point of the trajectory with respect to the point from which the ball is thrown is nearly ( $g = 10 \text{ ms}^{-2}$ )  
 (A)  $3.0 \text{ kg m}^2/\text{s}$  (B)  $3.46 \text{ kg m}^2/\text{s}$  (C)  $1.73 \text{ kg m}^2/\text{s}$  (D)  $6.0 \text{ kg m}^2/\text{s}$

Q.25

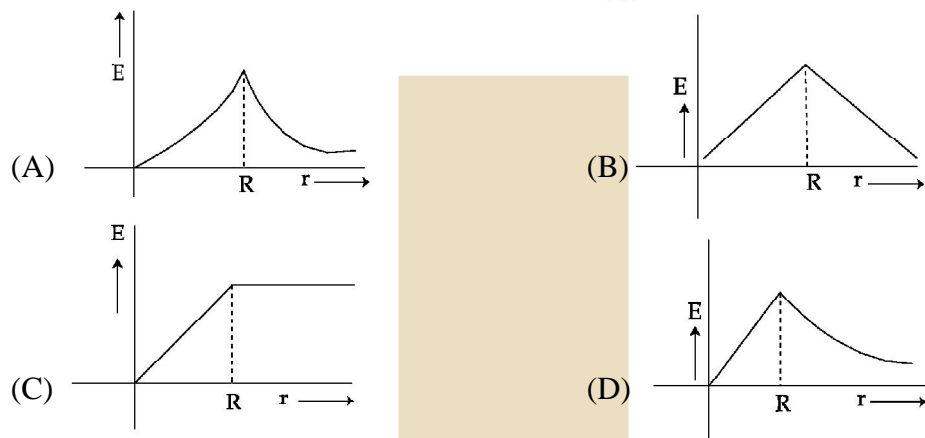
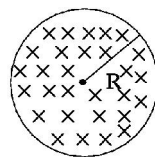


In an experiment for determining the gravitational acceleration  $g$  of a place with the help of a simple pendulum, the measured time period square is plotted against the string length of the pendulum in the figure. What is the value of  $g$  at the place ?

- (A)  $9.91 \text{ m/s}^2$  (B)  $10.0 \text{ m/s}^2$  (C)  $9.87 \text{ m/s}^2$  (D)  $9.81 \text{ m/s}^2$
- Q.26 An example of a perfect diamagnet is a superconductor. This implies that when a superconductor is put in a magnetic field of intensity  $B$ , the magnetic field  $B_s$  inside the superconductor will be such that :  
 (A)  $B_s < B$  but  $B_s \neq 0$  (B)  $B_s = B$  (C)  $B_s = 0$  (D)  $B_s = -B$
- Q.27 The total length of a sonometer wire between fixed ends is  $110 \text{ cm}$ . Two bridges are placed to divide the length of wire in ratio  $6 : 3 : 2$ . The tension in the wire is  $400 \text{ N}$  and the mass per unit length is  $0.01 \text{ kg/m}$ . What is the minimum common frequency with which three parts can vibrate?  
 (A)  $166 \text{ Hz}$  (B)  $1000 \text{ Hz}$  (C)  $1100 \text{ Hz}$  (D)  $100 \text{ Hz}$
- Q.28 A black coloured solid sphere of radius  $R$  and mass  $M$  is inside a cavity with vacuum inside. The walls of the cavity are maintained at temperature  $T_0$ . The initial temperature of the sphere is  $3T_0$ . If the specific heat of the material of the sphere varies as  $\alpha T^3$  per unit mass with the temperature  $T$  of the sphere, where  $\alpha$  is a constant, then the time taken for the sphere to cool down to temperature  $2T_0$  will be ( $\sigma$  is Stefan Boltzmann constant)

(A)  $\frac{M\alpha}{16\pi R^2\sigma} \ln\left(\frac{16}{3}\right)$  (B)  $\frac{M\alpha}{16\pi R^2\sigma} \ln\left(\frac{3}{2}\right)$  (C)  $\frac{M\alpha}{4\pi R^2\sigma} \ln\left(\frac{16}{3}\right)$  (D)  $\frac{M\alpha}{4\pi R^2\sigma} \ln\left(\frac{3}{2}\right)$

- Q.29 Figure shows a circular area of radius  $R$  where a uniform magnetic field  $\vec{B}$  is going into the plane of paper and increasing in magnitude at a constant rate. In that case, which of the following graphs, drawn schematically, correctly shows the variation of the induced electric field  $E(r)$ ?



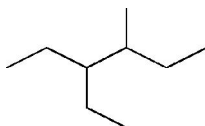
- Q.30 The diameter of the objective lens of microscope makes an angle  $\beta$  at the focus of the microscope. Further, the medium between the object and the lens is an oil of refractive index  $n$ . Then the resolving power of the microscope.

- (A) Increases with decreasing value of  $n$       (B) Increases with decreasing value of  $\frac{1}{n \sin 2\beta}$
- (C) Increases with increasing value of  $\beta$       (D) Increases with increasing value of  $n \sin 2\beta$

- Q.31 For an ideal solution of two components A and B, which of the following is true?  
 (A)  $\Delta H_{\text{mixing}} < 0$  (zero)  
 (B) A–A, B–B and A–B interactions are identical  
 (C) A–B interaction is stronger than A–A and B–B interactions  
 (D)  $\Delta H_{\text{mixing}} > 0$  (zero)
- Q.32 Which of these statements is not true ?  
 (A) In aqueous solution, the  $\text{Ti}^+$  ion is much more stable than  $\text{Ti}(\text{III})$   
 (B)  $\text{LiAlH}_4$  is a versatile reducing agent in organic synthesis.  
 (C)  $\text{NO}^+$  is not isoelectronic with  $\text{O}_2$   
 (D) B is always covalent in its compounds
- Q.33 Nickel ( $Z = 28$ ) combines with a uninegative monodentate ligand to form a diamagnetic complex  $[\text{NiL}_4]^{2-}$ . The hybridization involved and the number of unpaired electrons present in the complex are respectively :  
 (A)  $sp^3$ , zero                      (B)  $sp^3$ , two                      (C)  $dsp^2$ , one                      (D)  $dsp^2$ , zero
- Q.34 For the reaction,  
 $3A + 2B \rightarrow C + D$   
 the differential rate law can be written as :  
 (A)  $-\frac{1}{3} \frac{d[A]}{dt} = \frac{d[C]}{dt} = k[A]^n [B]^m$                       (B)  $+\frac{1}{3} \frac{d[A]}{dt} = \frac{d[C]}{dt} = k[A]^n [B]^m$   
 (C)  $\frac{1}{3} \frac{d[A]}{dt} = \frac{d[C]}{dt} = k[A]^n [B]^m$                       (D)  $-\frac{d[A]}{dt} = \frac{d[C]}{dt} = k[A]^n [B]^m$
- Q.35 Which one of the following molecules is paramagnetic?  
 (A) NO                      (B)  $\text{O}_3$                       (C)  $\text{N}_2$                       (D) CO
- Q.36 Ionization energy of gaseous Na atoms is  $495.5 \text{ kJ mol}^{-1}$ . The lowest possible frequency of light that ionizes a sodium atom is ( $h = 6.626 \times 10^{-34} \text{ Js}$ ,  $N_A = 6.022 \times 10^{23} \text{ mol}^{-1}$ )  
 (A)  $3.15 \times 10^{15} \text{ s}^{-1}$                       (B)  $4.76 \times 10^{14} \text{ s}^{-1}$                       (C)  $1.24 \times 10^{15} \text{ s}^{-1}$                       (D)  $7.50 \times 10^4 \text{ s}^{-1}$
- Q.37 The major product formed when 1, 1, 1- trichloro-propane is treated with aqueous potassium hydroxide is:  
 (A) 2 - Propanol                      (B) Propionic acid                      (C) Propyne                      (D) 1 - Propanol
- Q.38 The final product formed when Methyl amine is treated with  $\text{NaNO}_2$  and  $\text{HCl}$  is :  
 (A) Methylcyanide                      (B) Methylalcohol                      (C) Nitromethane                      (D) Diazomethane
- Q.39 Which one of the following is an example of thermosetting polymers?  
 (A) Nylon 6, 6                      (B) Bakelite                      (C) Buna -N                      (D) Neoprene
- Q.40 Which one of the following has largest ionic radius ?  
 (A)  $\text{Li}^+$                       (B)  $\text{F}^-$                       (C)  $\text{O}_2^{2-}$                       (D)  $\text{B}^{3+}$
- Q.41 Consider the reaction :  
 $\text{H}_2\text{SO}_{3(\text{aq})} + \text{Sn}_{(\text{aq})}^{4+} + \text{H}_2\text{O}_{(\text{l})} \longrightarrow \text{Sn}_{(\text{aq})}^{2+} + \text{HSO}_{4(\text{aq})}^- + 3\text{H}_{(\text{aq})}^+$   
 Which of the following statements is correct?  
 (A)  $\text{H}_2\text{SO}_3$  is the reducing agent because it undergoes reduction  
 (B)  $\text{H}_2\text{SO}_3$  is the reducing agent because it undergoes oxidation  
 (C)  $\text{Sn}^{4+}$  is the reducing agent because it undergoes oxidation  
 (D)  $\text{Sn}^{4+}$  is the oxidizing agent because it undergoes oxidation



Q.42 The correct IUPAC name of the following compound is:



- (A) 3, 4 – ethylmethylhexane                      (B) 3 – ethyl – 4 – methylhexane  
(C) 4 – ethyl – 3 – methylhexane                (D) 4 – methyl – 3 – ethylhexane

Q.43 Which one of the following ores is known as Malachite?

- (A)  $\text{Cu}_2\text{O}$                       (B)  $\text{CuFeS}_2$                       (C)  $\text{Cu}_2\text{S}$                       (D)  $\text{Cu}(\text{OH})_2 \cdot \text{CuCO}_3$

Q.44 For the decomposition of the compound, represented as  $\text{NH}_2\text{COONH}_4(\text{s}) \rightleftharpoons 2\text{NH}_3(\text{g}) + \text{CO}_2(\text{g})$  the  $K_p = 2.9 \times 10^{-5} \text{ atm}^3$ . If the reaction is started with 1 mol of the compound, the total pressure at equilibrium would be:

- (A)  $38.8 \times 10^{-2} \text{ atm}$     (B)  $1.94 \times 10^{-2} \text{ atm}$     (C)  $5.82 \times 10^{-2} \text{ atm}$     (D)  $7.66 \times 10^{-2} \text{ atm}$

Q.45 The reason for double helical structure of DNA is the operation of:

- (A) Electrostatic attractions                      (B) Hydrogen bonding  
(C) Dipole - Dipole interactions                (D) van der Waals forces

Q.46 Amongst  $\text{LiCl}$ ,  $\text{RbCl}$ ,  $\text{BeCl}_2$  and  $\text{MgCl}_2$  the compounds with the greatest and the least ionic character, respectively are:

- (A)  $\text{RbCl}$  and  $\text{MgCl}_2$     (B)  $\text{LiCl}$  and  $\text{RbCl}$     (C)  $\text{MgCl}_2$  and  $\text{BeCl}_2$     (D)  $\text{RbCl}$  and  $\text{BeCl}_2$

Q.47 Which one of the following compounds will not be soluble in sodium bicarbonate?

- (A) Benzene sulphonic acid                      (B) Benzoic acid  
(C) o – Nitrophenol                                (D) 2, 4, 6 - Trinitrophenol

Q.48 Among the following organic acids, the acid present in rancid butter is:

- (A) Lactic acid                      (B) Acetic acid                      (C) Pyruvic acid                      (D) Butyric acid

Q.49 The total number of octahedral void(s) per atom present in a cubic close packed structure is :

- (A) 1                                      (B) 2                                      (C) 3                                      (D) 4

Q.50 The observed osmotic pressure for a 0.10 M solution of  $\text{Fe}(\text{NH}_4)_2(\text{SO}_4)_2$  at  $25^\circ\text{C}$  is 10.8 atm. The expected and experimental (observed) values of Van't Hoff factor (i) will be respectively: ( $R = 0.082 \text{ L atm K mol}^{-1}$ )

- (A) 3 and 5.42                      (B) 5 and 3.42                      (C) 4 and 4.00                      (D) 5 and 4.42

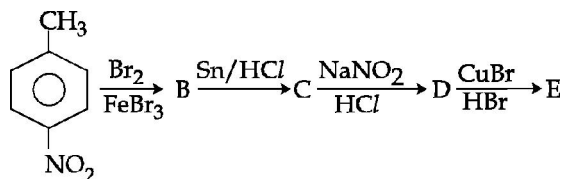
Q.51 Sulphur dioxide and oxygen were allowed to diffuse through a porous partition.  $20 \text{ dm}^3$  of  $\text{SO}_2$  diffuses through the porous partition in 60 seconds. The volume of  $\text{O}_2$  in  $\text{dm}^3$  which diffuses under the similar condition in 30 seconds will be (atomic mass of sulphur = 32 u) :

- (A) 28.2                                (B) 14.1                                (C) 7.09                                (D) 10.0

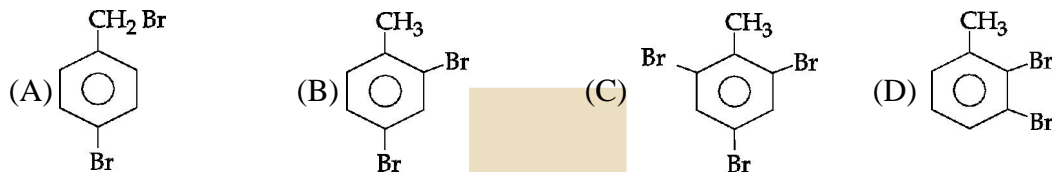
Q.52 An octahedral complex with molecular composition  $\text{M}_5\text{NH}_3 \cdot \text{Cl} \cdot \text{SO}_4$  has two isomers, A and B. The solution of A gives a white precipitate with  $\text{AgNO}_3$  solution and the solution of B gives white precipitate with  $\text{BaCl}_2$  solution. The type of isomerism exhibited by the complex is :

- (A) Geometrical isomerism                      (B) Coordinate isomerism  
(C) Ionization isomerism                        (D) Linkage isomerism

Q.53 In a set of reactions *p*-nitrotoluene yielded a product E



The product E would be:



Q.54 How many electrons are involved in the following redox reaction ?



- (A) 3 (B) 4 (C) 5 (D) 6

Q.55 Zirconium phosphate [ $\text{Zr}_3(\text{PO}_4)_4$ ] dissociates into three zirconium cations of charge + 4 and four phosphate anions of charge - 3. If molar solubility of zirconium phosphate is denoted by S and its solubility product by  $K_{sp}$ , then which of the following relationship between S and  $K_{sp}$  is correct?

- (A)  $S = \{K_{sp}/144\}^{1/7}$  (B)  $S = \{K_{sp}/(6912)^{1/7}$  (C)  $S = (K_{sp}/6912)^{1/7}$  (D)  $S = \{K_{sp}/6912\}^7$

Q.56 Choose the correct statement with respect to the vapour pressure of a liquid among the following:

- (A) Increases linearly with increasing temperature  
 (B) Decreases non-linearly with increasing temperature  
 (C) Decreases linearly with increasing temperature  
 (D) Increases non-linearly with increasing temperature

Q.57 Amongst the following, identify the species with an atom in +6 oxidation state :

- (A)  $[\text{MnO}_4]^-$  (B)  $[\text{Cr}(\text{CN})_6]^{3-}$  (C)  $\text{Cr}_2\text{O}_3$  (D)  $\text{CrO}_2\text{Cl}_2$

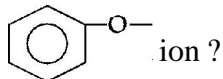
Q.58 Example of a three-dimensional silicate is:

- (A) Beryls (B) Zeolites (C) Feldspars (D) Ultramarines

Q.59 Williamson synthesis of ether is an example of:

- (A) Nucleophilic addition (B) Electrophilic addition  
 (C) Nucleophilic substitution (D) Electrophilic substitution

Q.60 Which one of the following substituents at *para*-position is most effective in stabilizing the phenoxide



- (A)  $-\text{CH}_3$  (B)  $-\text{CH}_2\text{OH}$  (C)  $-\text{OCH}_3$  (D)  $-\text{COCH}_3$



- Q.61 The circumcentre of a triangle lies at the origin and its centroid is the midpoint of the line segment joining the points  $(a^2 + 1, a^2 + 1)$  and  $2a, -1a$ ,  $a \neq 0$ . Then for any  $a$ , the orthocentre of this triangle lies on the line:
- (A)  $y - (a^2 + 1)x = 0$   
 (B)  $y + x = 0$   
 (C)  $(a - 1)^2x - (a + 1)^2y = 0$   
 (D)  $y - 2a = 0$
- Q.62 Let  $f(n) = \left[ \frac{1}{3} + \frac{3}{100} \right]^n$ , where  $[n]$  denotes the greatest integer less than or equal to  $n$ . Then  $\sum_{n=1}^{56} f(n)$  is equal to:
- (A) 56 (B) 1399 (C) 689 (D) 1287
- Q.63 If  $\vec{x} = 3\hat{i} - 6\hat{j} - \hat{k}$ ,  $\vec{y} = \hat{i} + 4\hat{j} - 3\hat{k}$  and  $\vec{z} = 3\hat{i} - 4\hat{j} - 12\hat{k}$  then the magnitude of the projection of  $\vec{x} \times \vec{y}$  on  $\vec{z}$  is:
- (A) 13 (B) 14 (C) 15 (D) 12
- Q.64 The equation of the circle described on the chord  $3x + y + 5$  of the circle  $x^2 + y^2 = 16$  as diameter is:
- (A)  $x^2 + y^2 + 3x + y - 2 = 0$  (B)  $x^2 + y^2 + 3x + y - 22 = 0$   
 (C)  $x^2 + y^2 + 3x + y + 1 = 0$  (D)  $x^2 + y^2 + 3x + y - 11 = 0$
- Q.65 The principal value of  $\tan^{-1}\left(\cot\frac{43\pi}{4}\right)$  is:
- (A)  $-\frac{3\pi}{4}$  (B)  $\frac{\pi}{4}$  (C)  $-\frac{\pi}{4}$  (D)  $\frac{3\pi}{4}$
- Q.66 Let A and E be any two events with positive probabilities:  
**Statement-1:**  $P(E/A) \geq P(A/E)P(E)$   
**Statement-2:**  $P(A/E) \geq P(A \cap E)$ .
- (A) Both the statements are false (B) Both the statements are true  
 (C) Statement-1 is false, statement-2 is true. (D) Statement-1 is true, statement-2 is false.
- Q.67 If a line L is perpendicular to the line  $5x - y = 1$ , and the area of the triangle formed by the line L and the coordinate axes is 5, then the distance of line L from the line  $x + 5y = 0$  is:
- (A)  $\frac{7}{\sqrt{5}}$  (B)  $\frac{5}{\sqrt{13}}$  (C)  $\frac{7}{\sqrt{13}}$  (D)  $\frac{5}{\sqrt{7}}$
- Q.68 Let A and B be any two  $3 \times 3$  matrices. If A is symmetric and B is skewsymmetric, then the matrix  $AB - BA$  is:
- (A) neither symmetric nor skewsymmetric (B) skewsymmetric  
 (C) symmetric (D) I or  $-I$ , where I is an identity matrix.

Q.69 If the angle between the line  $2(x + 1) = y = z + 4$  and the plane  $2x - y + \sqrt{\lambda}z + 4 = 0$  is  $\frac{\pi}{6}$ , then the value of  $\lambda$  is:-

- (A)  $\frac{45}{11}$                       (B)  $\frac{135}{7}$                       (C)  $\frac{45}{7}$                       (D)  $\frac{135}{11}$

Q.70 If  $\frac{dy}{dx} + y \tan x = \sin 2x$  and  $y(0) = 1$ , then  $y(\pi)$  is equal to :

- (A) 5                              (B) -1                              (C) 1                              (D) -5

Q.71 Two women and some men participated in a chess tournament in which every participant played two games with each of the other participants. If the number of games that the men played between themselves exceeds the number of games that the men played with the women by 66, then the number of men who participated in the tournament lies in the interval :

- (A) (14, 17)                      (B) [8, 9]                      (C) (11, 13]                      (D) [10, 12)

Q.72 The number of terms in an A.P. is even; the sum of the odd terms in it is 24 and that the even terms is 30. If the last term exceeds the first term by  $10\frac{1}{2}$ , then the number of terms in the A.P. is :

- (A) 8                              (B) 4                              (C) 16                              (D) 12

Q.73 If  $\Delta_r = \begin{vmatrix} r & 2r-1 & 3r-2 \\ \frac{n}{2} & n-1 & n \\ \frac{1}{2}n(n-1) & (n-1)^2 & \frac{1}{2}(n-1)(3n+4) \end{vmatrix}$  then the value of  $\sum_{r=1}^{n-1} \Delta_r$  :

- (A) depends only on n                              (B) is independent of both a and n  
 (C) depends only on a                              (D) depends both on a and n

Q.74 If non-zero real number b and c are such that  $\min f(x) > \max g(x)$ , where  $f(x) = x^2 + 2bx + 2c^2$  and  $g(x) = -x^2 - 2cx + b^2$  ( $x \in \mathbb{R}$ ) ;

then  $\left| \frac{c}{b} \right|$

- (A)  $\left[ \frac{1}{\sqrt{2}}, \sqrt{2} \right]$                       (B)  $(\sqrt{2}, \infty)$                       (C)  $\left[ \frac{1}{2}, \frac{1}{\sqrt{2}} \right)$                       (D)  $\left( 0, \frac{1}{2} \right)$

- Q.75 Equation of the line of the shortest distance between the lines  $\frac{x}{1} = \frac{y}{-1} = \frac{z}{1}$  and  $\frac{x-1}{0} = \frac{y+1}{-2} = \frac{z}{1}$  is:
- (A)  $\frac{x-1}{1} = \frac{y+1}{-2} = \frac{z}{1}$  (B)  $\frac{x}{-2} = \frac{y}{1} = \frac{z}{2}$  (C)  $\frac{x}{1} = \frac{y}{-1} = \frac{z}{-2}$  (D)  $\frac{x-1}{1} = \frac{y+1}{-1} = \frac{z}{-2}$
- Q.76 The coefficient of  $x^{1012}$  in the expansion of  $(1+x^n+x^{253})^{10}$ , (where  $n \leq 22$  is any positive integer), is:
- (A)  ${}^{253}C_4$  (B)  $4n$  (C) 1 (D)  ${}^{10}C_4$
- Q.77 The contrapositive of the statement "if I am not feeling well, then I will go to the doctor" is:
- (A) If I will go to the doctor, then I am not feeling well.  
 (B) If I will not go to the doctor, then I am feeling well.  
 (C) If I will go to the doctor, then I am feeling well.  
 (D) If I am feeling well, then I will not go to the doctor.
- Q.78 The function  $f(x) = |\sin 4x| + |\cos 2x|$ , is a periodic function with period:
- (A)  $2\pi$  (B)  $\pi/2$  (C)  $\pi$  (D)  $\pi/4$
- Q.79 Let  $f: \mathbb{R} \rightarrow \mathbb{R}$  be a function such that  $|f(x)| \leq x^2$ , for all  $x \in \mathbb{R}$ . Then, at  $x = 0$ ,  $f$  is:
- (A) Neither continuous nor differentiable (B) Differentiable but not differentiable  
 (C) Continuous as well as differentiable (D) Continuous but not differentiable
- Q.80 The area of the region above the  $x$ -axis bounded by the curve  $y = \tan x$ ,  $0 \leq x \leq \frac{\pi}{2}$  and the tangent to the curve at  $x = \frac{\pi}{4}$  is:
- (A)  $\frac{1}{2}(1 + \log 2)$  (B)  $\frac{1}{2}\left(\log 2 + \frac{1}{2}\right)$  (C)  $\frac{1}{2}(1 - \log 2)$  (D)  $\frac{1}{2}\left(\log 2 - \frac{1}{2}\right)$
- Q.81 If  $m$  is a non-zero number and  $\int \frac{x^{5m-1} + 2x^{4m-1}}{(x^{2m} + x^m + 1)^3} dx = f(x) + c$ , then  $f(x)$  is:
- (A)  $\frac{x^{5m}}{2m(x^{2m} + x^m + 1)^2}$  (B)  $\frac{x^{4m}}{2m(x^{2m} + x^m + 1)^2}$   
 (C)  $\frac{2m(x^{5m} + x^{4m})}{(x^{2m} + x^m + 1)^2}$  (D)  $\frac{(x^{5m} - x^{4m})}{2m(x^{2m} + x^m + 1)^2}$

Q.82 Let  $\bar{x}$ ,  $M$  and  $\sigma^2$  be respectively the mean, mode and variance of  $n$  observations  $x_1, x_2, \dots, x_n$  and  $d_i = -x_i - a$ ,  $i = 1, 2, \dots, n$ , where  $a$  is any number.

**Statement-1:** Variance of  $d_1, d_2, \dots, d_n$  is  $\sigma^2$ .

**Statement-2:** Mean and mode of  $d_1, d_2, \dots, d_n$  are  $-\bar{x} - a$  and  $-M - a$ , respectively

- (A) Statement-1 and statement-2 is both true.  
 (B) Statement-1 and statement-2 is both false.  
 (C) Statement-1 is true and statement-2 is false.  
 (D) Statement-1 is false and statement-2 is true.

Q.83 Let function  $F$  be defined as

$F(x) = \int_1^x \frac{e^t}{t} dt$ ,  $x > 0$  then the value of the integral  $\int_1^{x+a} \frac{e^t}{t+a} dt$ , where  $x > 0$ , is :-

- (A)  $e^a [F(x+a) - F(1+a)]$                       (B)  $e^{-a} [F(x+a) - F(1+a)]$   
 (C)  $e^{-a} [F(x+a) - F(a)]$                       (D)  $e^a [F(x) - F(1+a)]$

Q.84 If the function  $f(x) = \begin{cases} \frac{\sqrt{2 + \cos x} - 1}{(\pi - x)^2}, & x \neq \pi \\ k, & x = \pi \end{cases}$  is continuous at  $x = \pi$ , then  $k$  equals:-

- (A)  $\frac{1}{4}$                       (B)  $\frac{1}{2}$                       (C) 2                      (D) 0

Q.85 For all complex numbers  $z$  of the form  $1 + i\alpha$ ,  $\alpha \in \mathbb{R}$ , If  $z^2 = x + iy$ , then:-

- (A)  $y^2 - 4x + 2 = 0$     (B)  $y^2 - 4x + 4 = 0$     (C)  $y^2 + 4x - 4 = 0$     (D)  $y^2 + 4x + 2 = 0$

Q.86 If the volume of a spherical ball is increasing at the rate of  $4\pi$  cc/sec, then the rate of increase of its radius (in cm/sec), when the volume is  $288\pi$  cc, then the rate of increase of its radius (in cm/sec), when the volume is  $288\pi$  cc, is :

- (A)  $\frac{1}{24}$                       (B)  $\frac{1}{6}$                       (C)  $\frac{1}{9}$                       (D)  $\frac{1}{36}$

Q.87 The equation  $\sqrt{3x^2 + x + 5} = x - 3$ , where  $x$  is real, has:

- (A) no solution                      (B) exactly two solutions  
 (C) exactly one solution                      (D) exactly four solutions

Q.88 The tangent at an extremity (in the first quadrant) of latus rectum of the hyperbola  $\frac{x^2}{4} - \frac{y^2}{5} = 1$ , meets x-axis and y-axis at A and B respectively. Then  $(OA)^2 - (OB)^2$ , where O is the origin, equals:

- (A)  $\frac{16}{9}$                       (B)  $-\frac{20}{9}$                       (C)  $-\frac{4}{3}$                       (D) 4

Q.89 A chord is drawn through the focus of the parabola  $y^2 = 6x$  such that its distance from the vertex of this parabola is  $\frac{\sqrt{5}}{2}$ , then its slope can be:

- (A)  $\frac{\sqrt{3}}{2}$                       (B)  $\frac{\sqrt{5}}{2}$                       (C)  $\frac{2}{\sqrt{3}}$                       (D)  $\frac{2}{\sqrt{5}}$

Q.90 Let  $f : \mathbb{R} \rightarrow \mathbb{R}$  be defined by  $f(x) = \frac{|x|-1}{|x|+1}$  then f is :

- (A) one-one but not onto                      (B) neither one-one nor onto  
(C) both one-one and onto                      (D) onto but not one-one

JEE MAIN & AIEEE-2014 QUESTION OF ON LINE EXAMINATION										
ANSWER KEY										PAPER-4
Q.	1	2	3	4	5	6	7	8	9	10
A.	D	A	A	C	B	B	C	C	A	A
Q.	11	12	13	14	15	16	17	18	19	20
A.	C	BONUS	D	B	B	A	D	B	B	D
Q.	21	22	23	24	25	26	27	28	29	30
A.	D	A	B	A	C	C	B	A	D	D
Q.	31	32	33	34	35	36	37	38	39	40
A.	B	C	D	A	A	C	B	B	B	C
Q.	41	42	43	44	45	46	47	48	49	50
A.	A	B	D	C	B	D	C	D	A	D
Q.	51	52	53	54	55	56	57	58	59	60
A.	B	C	B	D	C	D	D	A	C	D
Q.	61	62	63	64	65	66	67	68	69	70
A.	C	B	B	D	C	B	B	C	C	D
Q.	71	72	73	74	75	76	77	78	79	80
A.	A	B	B	B	D	D	B	B	C	D
Q.	81	82	83	84	85	86	87	88	89	90
A.	B	A	B	A	C	D	A	B	B	B